

# INORGANIC COMPOUNDS

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## Low-Temperature Form of $\text{LiGaSi}_2\text{O}_6$

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### Abstract

The structure of the low-temperature form of gallium lithium metasilicate,  $\text{LiGaSi}_2\text{O}_6$ , has been determined from diffractometer data obtained at 273 K. The title structure is similar to that of pigeonite  $[(\text{Mg}_{0.39}\text{Fe}_{0.52}\text{Ca}_{0.09})\text{SiO}_3]$ .

### Comment

Solid-state  $^{69}\text{Ga}$  NMR spectra of the title compound, (I), indicate that there is a phase transition dependent on the temperature (Ohashi, Osawa, Sato & Onoda, 1995). Compound (I) is isostructural with  $(\text{Mg}_{0.39}\text{Fe}_{0.52}\text{Ca}_{0.09})\text{SiO}_3$  (Morimoto & Guven, 1970) and the low-temperature forms of  $\text{LiFeSi}_2\text{O}_6$  and  $\text{LiCrSi}_2\text{O}_6$  (Behruzi, Hahn, Prewitt & Baldwin, 1984). Li is coordinated by seven O atoms and Ga is coordinated by six O atoms. There are two crystallographically different kinds of single silicate chain in the structure. This low-temperature form ( $P2_1/c$ ) transforms to the high temperature form ( $C2/c$ ) at 285 K (Sato, Osawa & Ohashi, 1994).

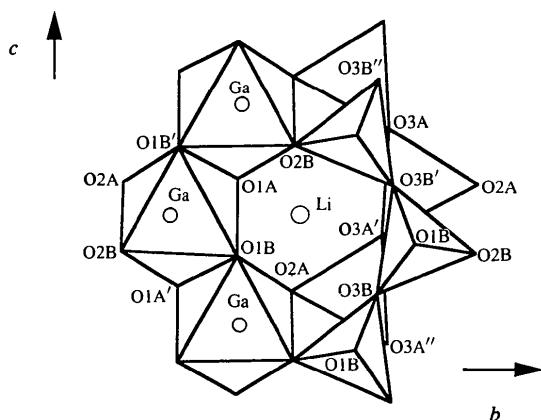


Fig. 1. Projection of the structure of  $\text{LiGaSi}_2\text{O}_6$ . The nomenclature of the O atoms is that used by Morimoto & Guven (1970).

### Experimental

The title compound was prepared according to the procedure described by Ohashi, Osawa & Sato (1989).

### Crystal data

$\text{LiGaSi}_2\text{O}_6$	Mo $K\alpha$ radiation
$M_r = 228.83$	$\lambda = 0.71073 \text{ \AA}$
Monoclinic	Cell parameters from 22 reflections
$P2_1/c$	$\theta = 27.7\text{--}32.0^\circ$
$a = 9.5394(4) \text{ \AA}$	$\mu = 7.32 \text{ mm}^{-1}$
$b = 8.5756(2) \text{ \AA}$	$T = 273 \text{ K}$
$c = 5.2508(3) \text{ \AA}$	Prism
$\beta = 110.124(2)^\circ$	$0.28 \times 0.14 \times 0.13 \text{ mm}$
$V = 403.32(3) \text{ \AA}^3$	Colourless
$Z = 4$	
$D_x = 3.77 \text{ Mg m}^{-3}$	

### Data collection

Enraf–Nonius CAD-4	$\theta_{\max} = 40^\circ$
diffractometer	$h = 0 \rightarrow 16$
$\omega/2\theta$ scans	$k = 0 \rightarrow 15$
Absorption correction:	$l = -9 \rightarrow 9$
by integration from crystal shape	3 standard reflections monitored every 100 reflections for orientation and every 240 min for intensity
$T_{\min} = 0.343$ , $T_{\max} = 0.439$	intensity decay: 1.2%
2605 measured reflections	
2605 independent reflections	
2127 observed reflections	
$[I > 1.5\sigma(I)]$	

### Refinement

Refinement on $F$	isotropic secondary extinction
$R = 0.041$	Extinction coefficient:
$wR = 0.043$	$3.16 \times 10^{-6}$
$S = 1.4$	Atomic scattering factors
2127 reflections	from International Tables for X-ray Crystallography (1974, Vol. IV)
92 parameters	
Unit weights applied	
$(\Delta/\sigma)_{\max} < 0.005$	
$\Delta\rho_{\max} = 2.0 \text{ e \AA}^{-3}$	
$\Delta\rho_{\min} = -2.2 \text{ e \AA}^{-3}$	

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{eq}}$
Ga	0.25033 (3)	0.65098 (3)	0.23963 (5)	0.00407 (4)
Li	0.2479 (9)	0.0125 (9)	0.237 (1)	0.016 (1)
SiA	0.04726 (8)	0.34077 (9)	0.2683 (1)	0.0041 (1)
SiB	0.54799 (8)	0.84060 (9)	0.2525 (1)	0.0040 (1)
O1A	0.8640 (2)	0.3326 (2)	0.1534 (3)	0.0048 (3)
O2A	0.1148 (2)	0.5121 (3)	0.3132 (4)	0.0075 (4)
O3A	0.1081 (2)	0.2623 (3)	0.5688 (4)	0.0070 (3)
O1B	0.3645 (2)	0.8336 (3)	0.1340 (4)	0.0052 (3)
O2B	0.6189 (2)	1.0085 (3)	0.3431 (4)	0.0076 (4)
O3B	0.6064 (2)	0.7307 (3)	0.5210 (4)	0.0086 (4)

Table 2. Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

SiA—O1A <sup>i</sup>	1.643 (2)	O2B <sup>iii</sup> —O3B <sup>iii</sup>	2.635 (3)
SiA—O2A	1.589 (2)	O2B <sup>iii</sup> —O3B	2.576 (3)
SiA—O3A <sup>i</sup>	1.628 (2)	O3B <sup>iii</sup> —O3B	2.646 (3)
SiA—O3A <sup>ii</sup>	1.623 (2)	O1A <sup>v</sup> —O1A <sup>iv</sup>	2.983 (3)
SiB—O1B	1.645 (2)	O1A <sup>v</sup> —O2A	2.857 (3)
SiB—O2B	1.592 (2)	O1A <sup>v</sup> —O1B	2.607 (3)
SiB—O3B <sup>iii</sup>	1.621 (3)	O1A <sup>v</sup> —O2B <sup>vii</sup>	2.713 (3)
SiB—O3B	1.626 (2)	O1A <sup>iv</sup> —O2A	2.757 (3)
Ga—O1A <sup>iv</sup>	2.086 (2)	O1A <sup>iv</sup> —O1B	2.770 (3)
Ga—O1A <sup>v</sup>	1.980 (2)	O1A <sup>iv</sup> —O1B <sup>vi</sup>	2.607 (3)
Ga—O1B	2.089 (2)	O2A <sup>vi</sup> —O1B <sup>vi</sup>	2.739 (3)
Ga—O1B <sup>vi</sup>	1.985 (2)	O2A <sup>vi</sup> —O2B <sup>vii</sup>	2.927 (3)
Ga—O2A	1.893 (2)	O1B <sup>vi</sup> —O1B <sup>vi</sup>	2.992 (3)
Ga—O2B <sup>vii</sup>	1.900 (2)	O1B <sup>vi</sup> —O2B <sup>vii</sup>	2.793 (3)
O1A <sup>i</sup> —O2A	2.723 (3)	O1B <sup>vi</sup> —O2B <sup>vii</sup>	2.900 (3)
O1A <sup>i</sup> —O3A <sup>i</sup>	2.655 (3)	Li—O1A <sup>vii</sup>	2.066 (8)
O1A <sup>i</sup> —O3A <sup>ii</sup>	2.645 (3)	Li—O1B <sup>vii</sup>	2.074 (8)
O2A <sup>vi</sup> —O3A <sup>i</sup>	2.540 (3)	Li—O2A <sup>vi</sup>	2.162 (6)
O2A <sup>vi</sup> —O3A <sup>ii</sup>	2.670 (3)	Li—O2B <sup>vi</sup>	2.138 (6)
O3A <sup>vi</sup> —O3A <sup>ii</sup>	2.634 (3)	Li—O3A <sup>ii</sup>	2.341 (7)
O1B <sup>vi</sup> —O2B	2.740 (3)	Li—O3B <sup>vii</sup>	2.928 (9)
O1B <sup>vi</sup> —O3B <sup>iii</sup>	2.631 (3)	Li—O3B <sup>vii</sup>	2.679 (7)
O1B <sup>vi</sup> —O3B	2.647 (2)		
O1A <sup>i</sup> —SiA—O2A	114.8 (1)	Ga—O1B—Ga <sup>iii</sup>	100.2 (1)
O1A <sup>i</sup> —SiA—O3A	108.5 (1)	O1A <sup>v</sup> —Ga—O1A <sup>iv</sup>	94.4 (1)
O1A <sup>i</sup> —SiA—O3A <sup>ii</sup>	108.2 (1)	O1A <sup>v</sup> —Ga—O2A	95.0 (1)
O2A—SiA—O3A	104.3 (1)	O1A <sup>v</sup> —Ga—O1B	79.6 (1)
O2A—SiA—O3A <sup>ii</sup>	112.5 (1)	O1A <sup>v</sup> —Ga—O1B <sup>vi</sup>	172.1 (1)
O3A—SiA—O3A <sup>ii</sup>	108.2 (1)	O1A <sup>v</sup> —Ga—O2B <sup>vii</sup>	88.7 (1)
O1B—SiB—O2B	115.7 (1)	O1A <sup>iv</sup> —Ga—O2A	87.6 (1)
O1B—SiB—O3B <sup>iii</sup>	107.3 (1)	O1A <sup>iv</sup> —Ga—O1B	83.1 (1)
O1B—SiB—O3B	108.0 (1)	O1A <sup>iv</sup> —Ga—O1B <sup>vi</sup>	79.6 (1)
O2B—SiB—O3B <sup>iii</sup>	110.2 (1)	O1A <sup>iv</sup> —Ga—O2B <sup>vii</sup>	170.7 (1)
O2B—SiB—O3B	106.3 (1)	O2A—Ga—O1B	168.9 (1)
O3B <sup>iii</sup> —SiB—O3B	109.1 (1)	O2A—Ga—O1B <sup>vi</sup>	89.9 (1)
SiA—O3A—SiA <sup>x</sup>	139.8 (1)	O2A—Ga—O2B <sup>vii</sup>	101.0 (1)
O3A <sup>vi</sup> —O3A <sup>ii</sup> —O3A <sup>x</sup>	170.8 (1)	O1B <sup>vi</sup> —Ga—O1B <sup>vi</sup>	94.5 (1)
Ga—O1A <sup>iv</sup> —Ga <sup>vii</sup>	100.5 (1)	O1B <sup>vi</sup> —Ga—O2B <sup>vii</sup>	88.8 (1)
SiB—O3B <sup>iii</sup> —SiB <sup>iii</sup>	139.9 (2)	O1B <sup>vii</sup> —Ga—O2B <sup>vii</sup>	96.5 (1)
O3B <sup>iii</sup> —O3B <sup>vi</sup> —O3B <sup>vii</sup>	165.7 (1)		

Symmetry codes: (i)  $x - 1, y, z$ ; (ii)  $x, \frac{1}{2} - y, z - \frac{1}{2}$ ; (iii)  $x, \frac{3}{2} - y, z - \frac{1}{2}$ ; (iv)  $1 - x, \frac{1}{2} + y, \frac{1}{2} - z$ ; (v)  $1 - x, 1 - y, -z$ ; (vi)  $x, \frac{3}{2} - y, \frac{1}{2} + z$ ; (vii)  $1 - x, y - \frac{1}{2}, \frac{1}{2} - z$ ; (viii)  $x, y - 1, z$ ; (ix)  $1 - x, 1 - y, 1 - z$ ; (x)  $x, \frac{1}{2} - y, \frac{1}{2} + z$ ; (xi)  $x, y, z - 1$ .

The structure was solved by direct methods with *MULTAN11/82* (Main *et al.*, 1982). All calculations were performed with the *SDP* program system (B. A. Frenz & Associates Inc., 1985).

Lists of structure factors and anisotropic displacement parameters have been deposited with the IUCr (Reference: OH1080). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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## TaOI<sub>2</sub>: Eine zentrosymmetrische Variante der NbOI<sub>2</sub>-Struktur

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## Abstract

The crystal structure of tantalum(IV) oxide diiodide was determined by single-crystal X-ray diffraction. TaOI<sub>2</sub> contains sheets of distorted TaO<sub>2</sub>I<sub>4</sub> octahedra which share *trans* I—I edges and *trans* O corners (centrosymmetric MoOCl<sub>2</sub> type). The Ta atoms are displaced from the centres of the octahedra forming Ta···Ta pairs [3.100 (2)  $\text{\AA}$ ]. The structure is closely related to the structure of NbOI<sub>2</sub>, which was reported as the non-centrosymmetric NbOCl<sub>2</sub> type.

## Kommentar

TaOI<sub>2</sub> kristallisiert im MoOCl<sub>2</sub>-Typ (Schäfer & Schnering, 1964). Die Ta-Atome sind verzerrt oktaedrisch von je zwei *trans*-ständigen O-Atomen und vier I-Atomen in der Oktaederbasis umgeben. Über die I-Atome sind diese Oktaeder zu linearen Strängen verbunden. Die Verzerrung der Oktaeder resultiert aus dem paarweisen Zusammenrücken der Ta-Atome [ $d(\text{Ta}—\text{Ta}) = 3.100 (2) \text{\AA}$ ], so daß genauer von Ketten aus *trans*-kantenverknüpften Ta<sub>2</sub>O<sub>4</sub>I<sub>6</sub>-Doppeloktaedern gesprochen werden muß. Die gleichzeitige Eckerverknüpfung der Oktaeder über gemeinsame O-Atome führt zu Schichten, die zusammenfassend als  $\infty[\text{Ta}_2\text{O}_{4/2}\text{I}_{2}\text{I}_{4/2}]$  formuliert werden können. Die Oberfläche einer solchen Schicht besteht aus einem leicht gewellten, angenähert quadratischen Netz von I-Atomen. Die Stapelung der Schichten entlang der kristallographischen *a*-Achse erfolgt in der Weise, daß die Iodatome der benachbarten Schicht über den Mittelpunkten dieser Quadrate liegen.

Der strukturelle Aufbau von TaOI<sub>2</sub> (Raumgruppe *C2/m*) lehnt sich eng die im NbOCl<sub>2</sub>-Typ kristallisierenden Nioboxiddihalogenide (Schäfer & Schnering, 1964) an. In einer detaillierteren Arbeit über die Struktur von NbOI<sub>2</sub> (Rijnsdorp & Jellinek, 1978)